

# Synthesis and activity of penicillin G hydroxamic acid derivatives against *S. aureus*, *S. pneumoniae*, and intraphagocytic *L. monocytogenes*.

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## ABSTRACT

**Background:** Hydroxamic acid derivatives of beta-lactams (carboxyl group replaced by a CO(NH)R<sub>1</sub>O-R<sub>2</sub> moiety) are not ionizable at physiological pH, which could allow for improved penetration in phagocytic cells. We have synthesized new series of N-hydroxamic acid derivatives of penicillins and their activity against Gram-positive is maintained against both extracellular and intracellular bacteria.

**Methods:** Syntheses were made with the mixed anhydride method, with starting compounds prepared by modified Gabriel's and Delechâtel's procedures for O-alkylated and N-hydroxymethyl derivatives. MICs were determined by agar dilution and activity on intracellular L. m. evaluated on infected murine J774 macrophages (Chanteux et al., JAC, 52:6105-2003). The absence of conversion into Pen G in microbial and cell culture media was checked by HPLC.

**Results:** Structures were confirmed by <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, IR, and HR mass spectrometry. No release of free Pen G was detected. MICs (mg/L) are shown in Table.

Bacteria	PenG	Oxa	1	2	3	4	5	R <sub>1</sub> = iPr R <sub>2</sub> = CH <sub>3</sub>	R <sub>1</sub> = H R <sub>2</sub> = Bz	R <sub>1</sub> = H R <sub>2</sub> = CH <sub>3</sub>
S.a.(n=4)	0.03-0.25	0.25	3	2	3	4	5	1	1	1
S.p.(n=6)	≤ 0.03	0.08-0.25	0.06-0.25	0.5-2	0.06-0.125	< 0.03	2	1	1	1
L.m.	0.06									

1-4 are weak acids (pK approx. 9); 5 is not an acid.

On intracellular L. m., Pen G and 2 were both static at an extracellular concent. (Ce) of 6 × the MIC, but 1 was already static at Ce = 1 × the MIC.

**Conclusion:** Replacing the free carboxyl group of beta-lactams by a weak acid does not necessarily suppress antibacterial activity and may confer enhanced activity against intracellular bacteria at equipotent Ce.

The activity of  $\beta$ -lactam antibiotics is critically dependent on the presence of a free carboxyl function or of an equivalent acidic group at a strategic distance from the  $\beta$ -lactam ring in order to mimic a D-Ala-D-Ala moiety (1,2). This acid function, however, is responsible for the lack of cellular penetration and tissue accumulation of  $\beta$ -lactams (3).

Hydroxamic acid derivatives of beta-lactams keep the acidic character of the classical carboxyl-substituted beta-lactams, while being not ionizable at physiological pH, which could allow their improved penetration in eucaryotic cells.

## INTRODUCTION

The activity of  $\beta$ -lactam antibiotics is critically dependent on the presence of a free carboxyl function or of an equivalent acidic group at a strategic distance from the  $\beta$ -lactam ring in order to mimic a D-Ala-D-Ala moiety (1,2). This acid function, however, is responsible for the lack of cellular penetration and tissue accumulation of  $\beta$ -lactams (3).

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## METHODS

**Synthesis of N-hydroxybenzylpenicillin derivatives:** All compounds were prepared by the mixture anhydride method (4). They were characterized by infrared and nuclear magnetic resonance (<sup>1</sup>H NMR and <sup>13</sup>C NMR) spectroscopy, melting points and high-resolution mass spectrometry.

**Microbiological assays:** MICs were determined by standard microdilution techniques according to NCCLS guidelines using ATCC strains and clinical isolates of *S. aureus* and *S. pneumoniae*.

**Intracellular activity against *L. monocytogenes*:** Intracellular activity was testing in a model of J774 mouse macrophages over a 5 h period, following the procedure described in our previous publications (see 5 for example).

**Stability:** of the molecules was checked by HPLC in all conditions of biological assays.

## REFERENCES

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